

# Download Ph Problems And Solutions

Problem : What is the pH of a solution of 0.36 M HCl, 0.62 M NaOH, and 0.15 M HNO<sub>3</sub>? Hydrochloric acid and nitric acid are strong acids, and sodium hydroxide is a ...pH Problems FORMULAS ... B. Calculate the concentrations of H<sup>+</sup> and OH<sup>-</sup> in the following solutions: 1) lemon juice, pH = 2.30 5) blood, pH = 7.40 2) ...This chemistry video tutorial explains how to calculate the pH of a buffer solution using the henderson hasselbalch equation. It explains the concept ...Acids & Bases. This video shows you how to calculate the pOH of a solution when given the [OH<sup>-</sup>] concentration. From the pOH you can then solve for the pH ...